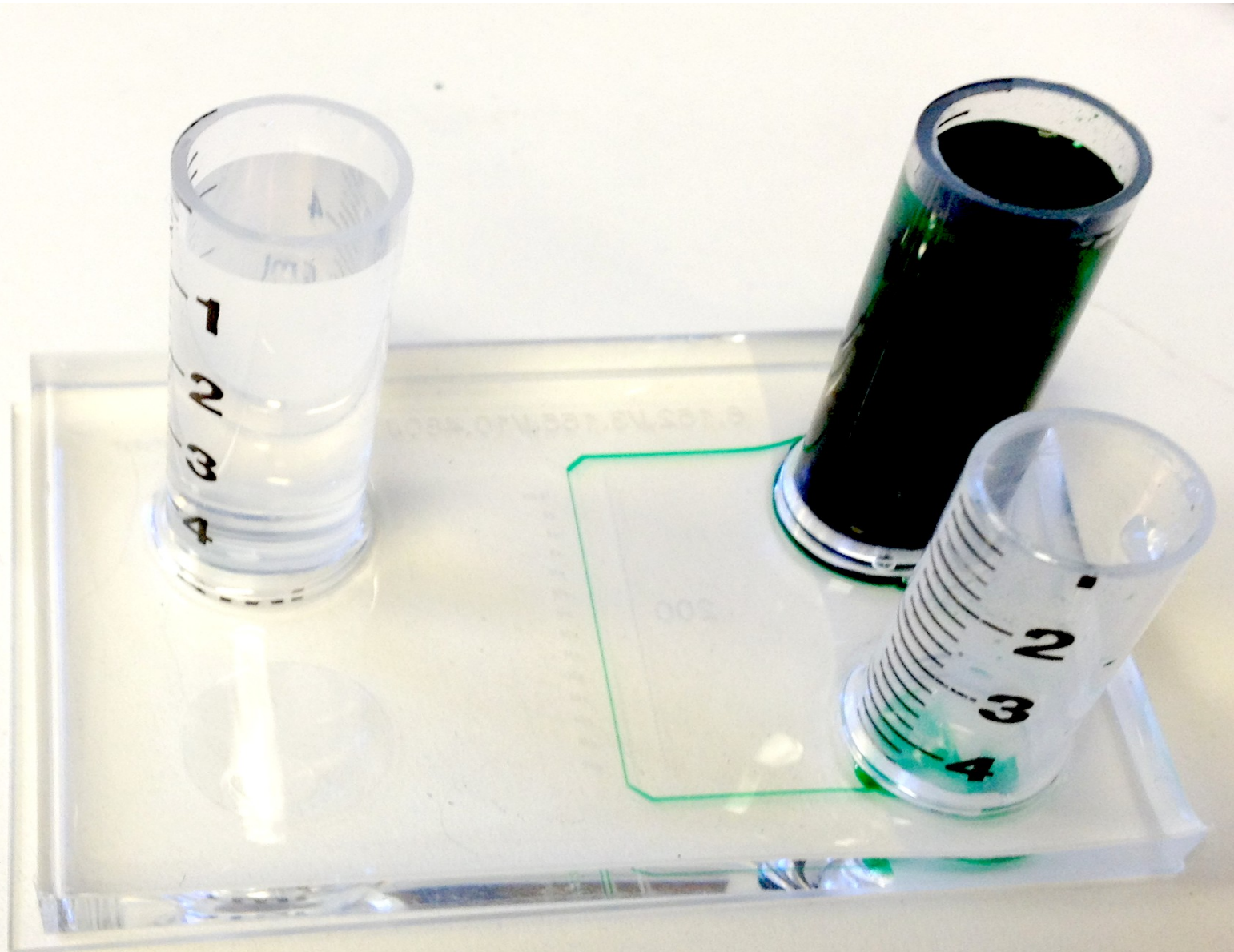


Diffusion in Microfluidic Mixers

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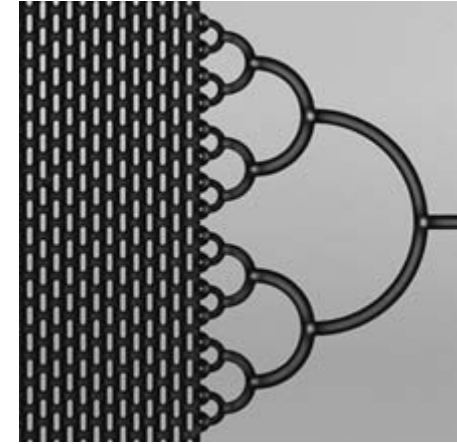
Experimenters:

Sam Bader,

Juan Hernandez,

William Baysinger

Why Microfluidics?

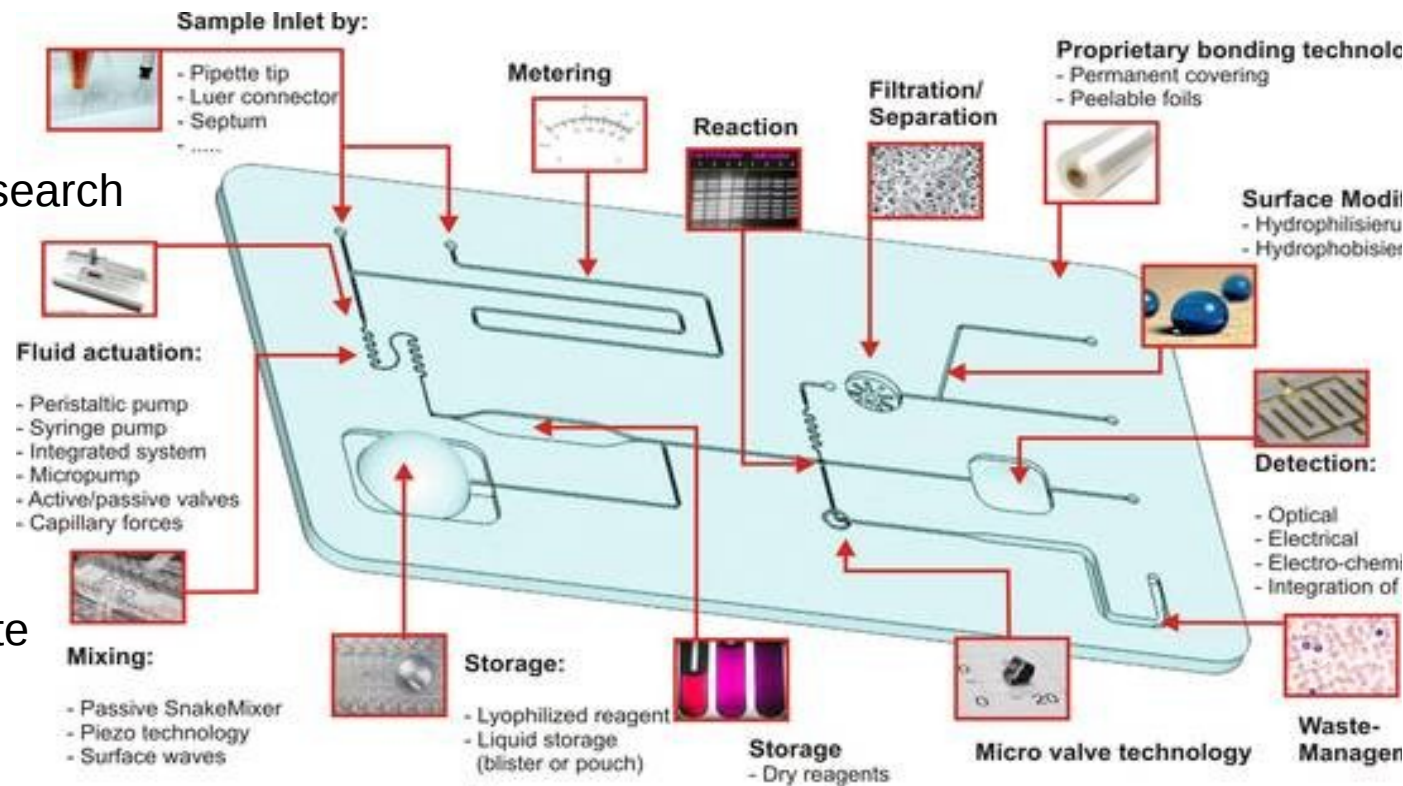


- **Applications of microfluidics:**

- Printing
- Chemical and medical research
- Cooling electronics

- **Advantages:**

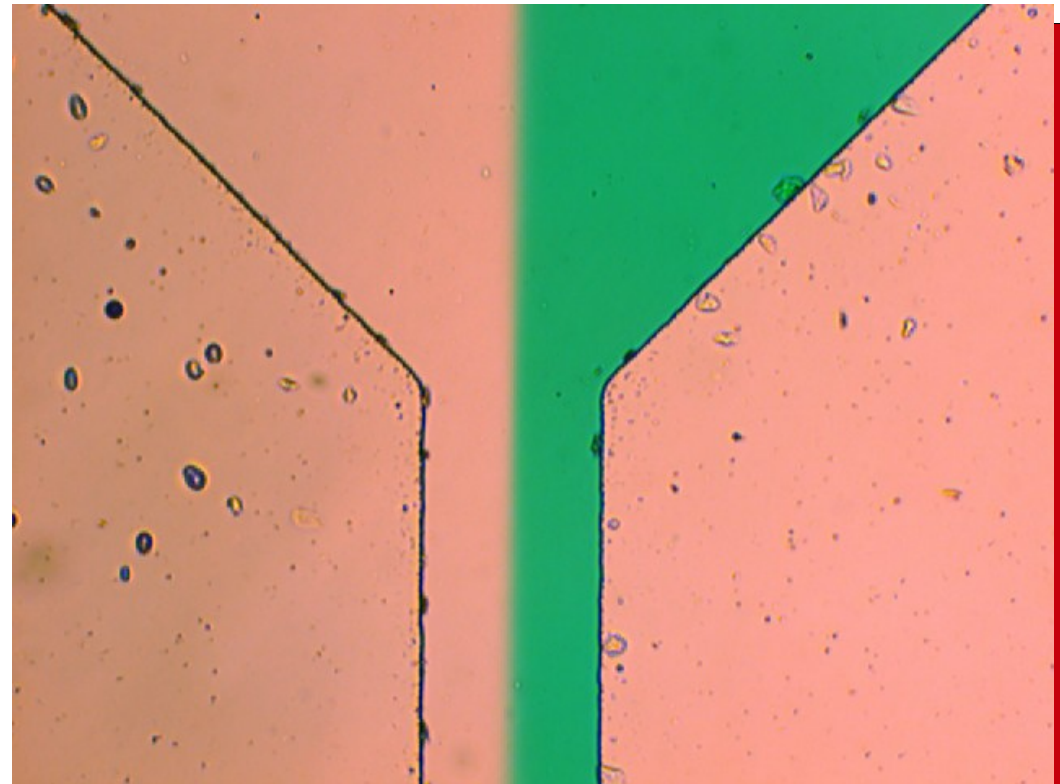
- Compact
- Easy to integrate, automate
- Use smaller samples
- Lab-on-a-chip



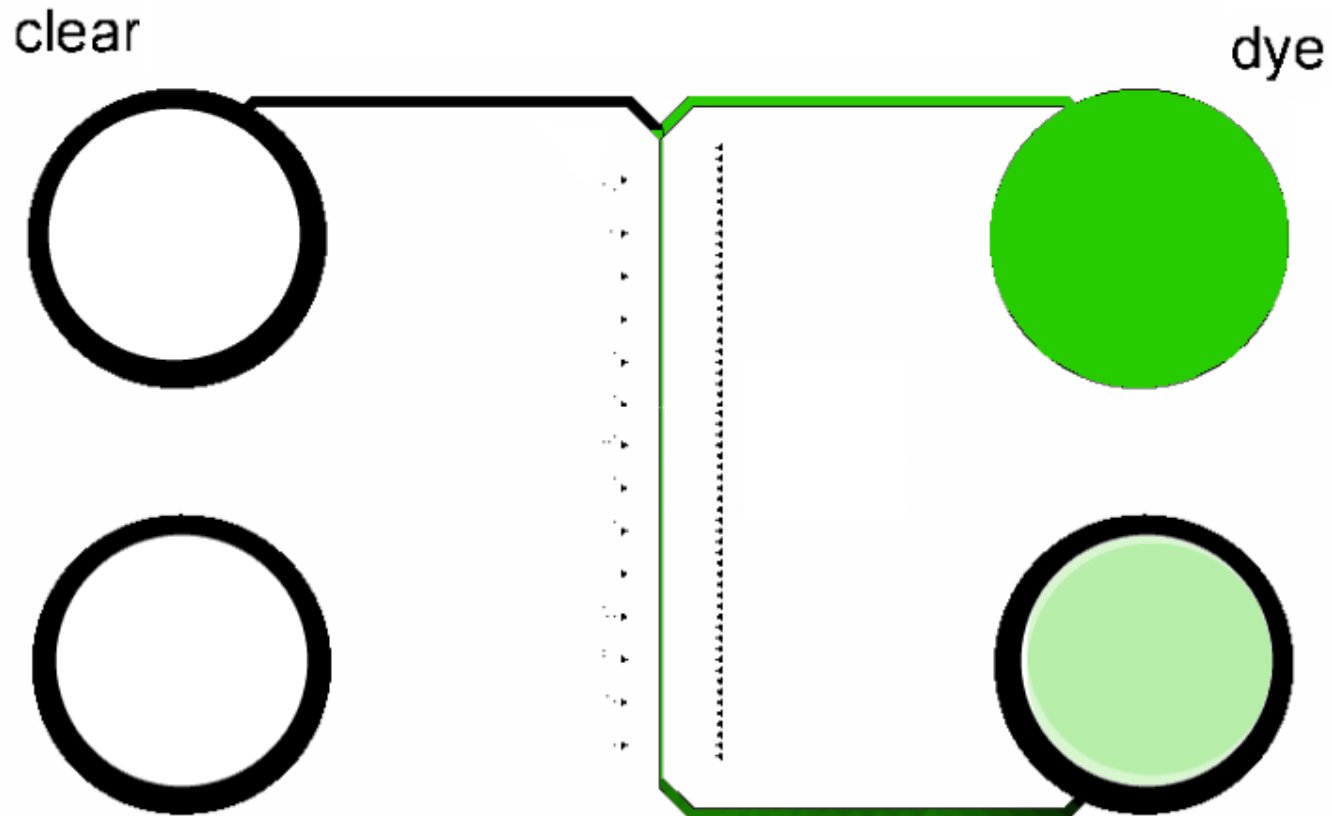
Diffusion is Vital

- ***Micro is different:***

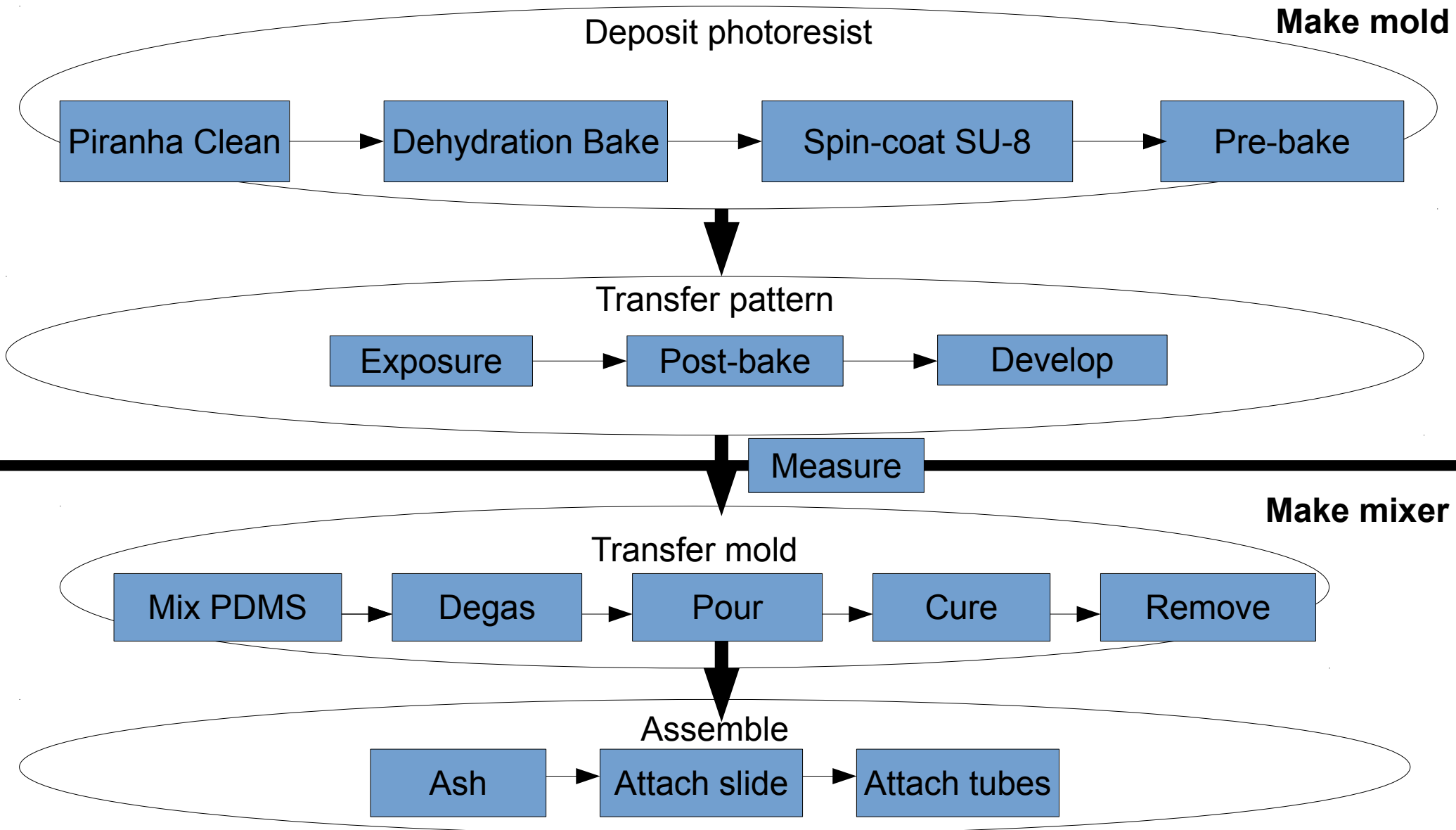
- Small Reynold's number, and complex flow patterns
- Scaling: surface and entrance effects important
- In absence of turbulence,
mixing occurs by diffusion



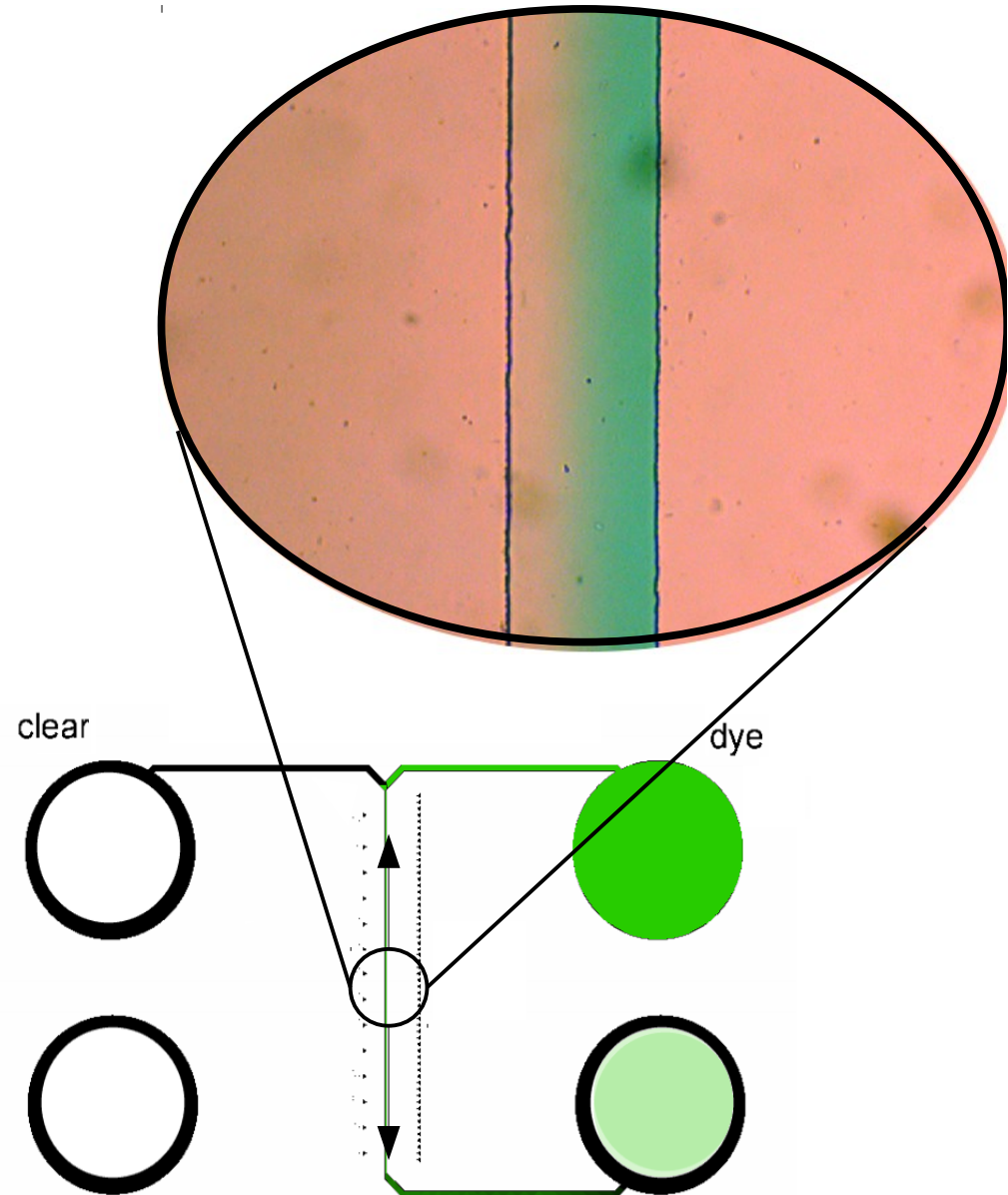
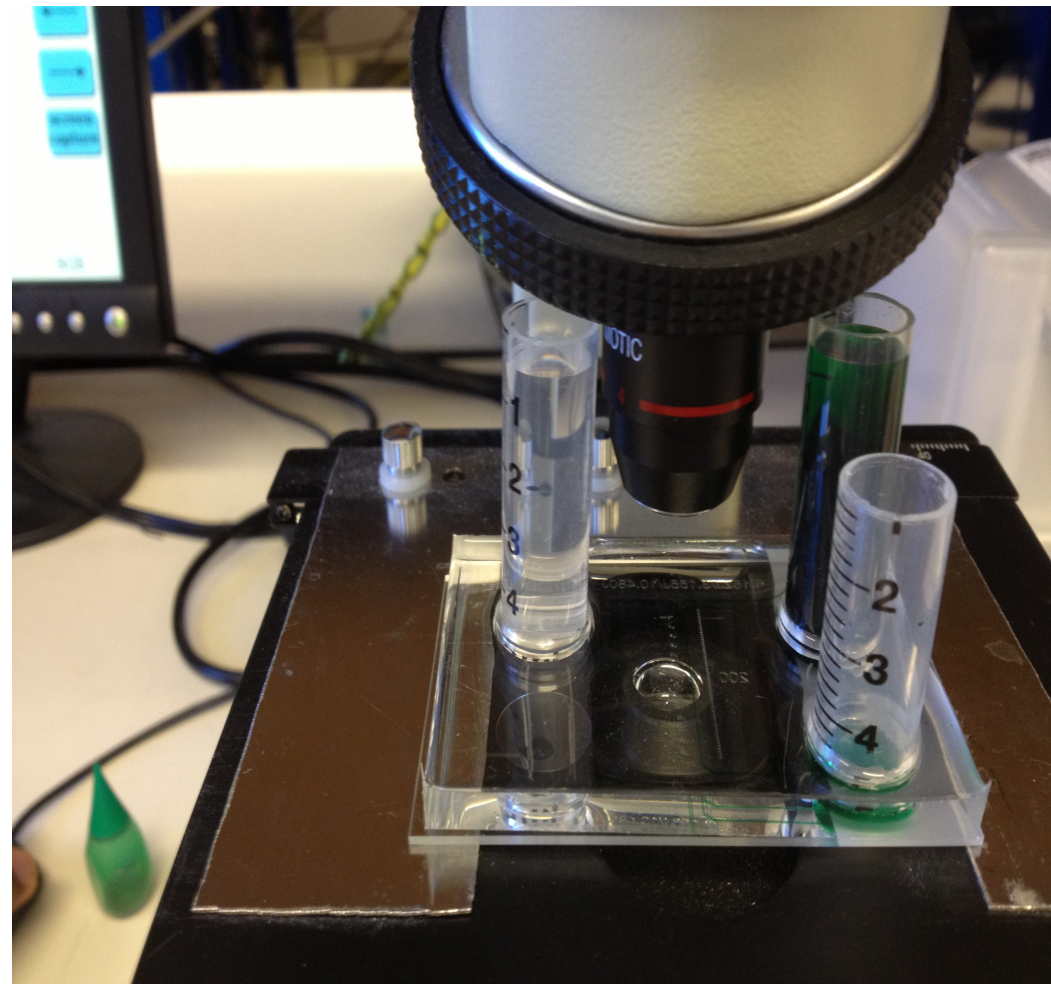
Characterize the diffusive mixing



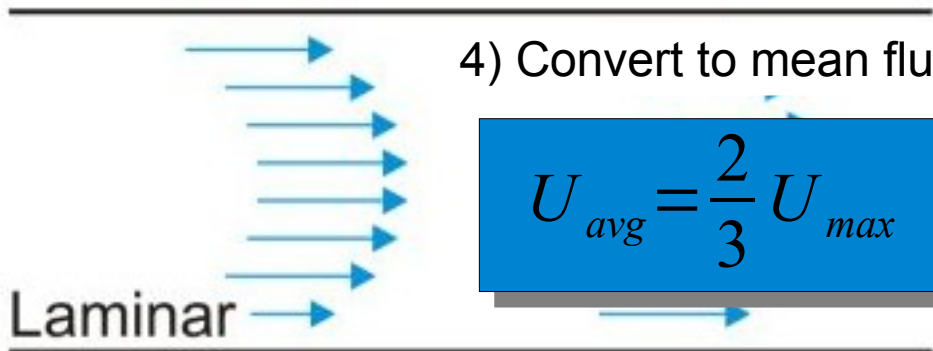
Fabrication



Experiment and Observe



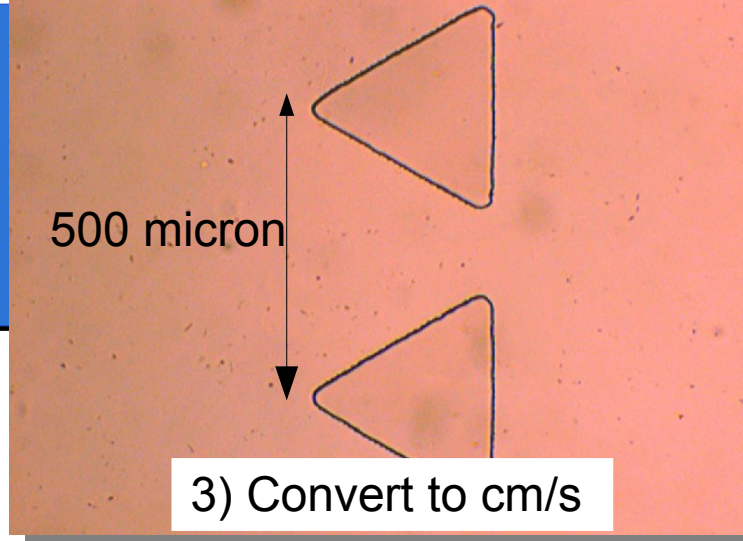
Determine fluid velocity



4) Convert to mean fluid velocity

$$U_{avg} = \frac{2}{3} U_{max}$$

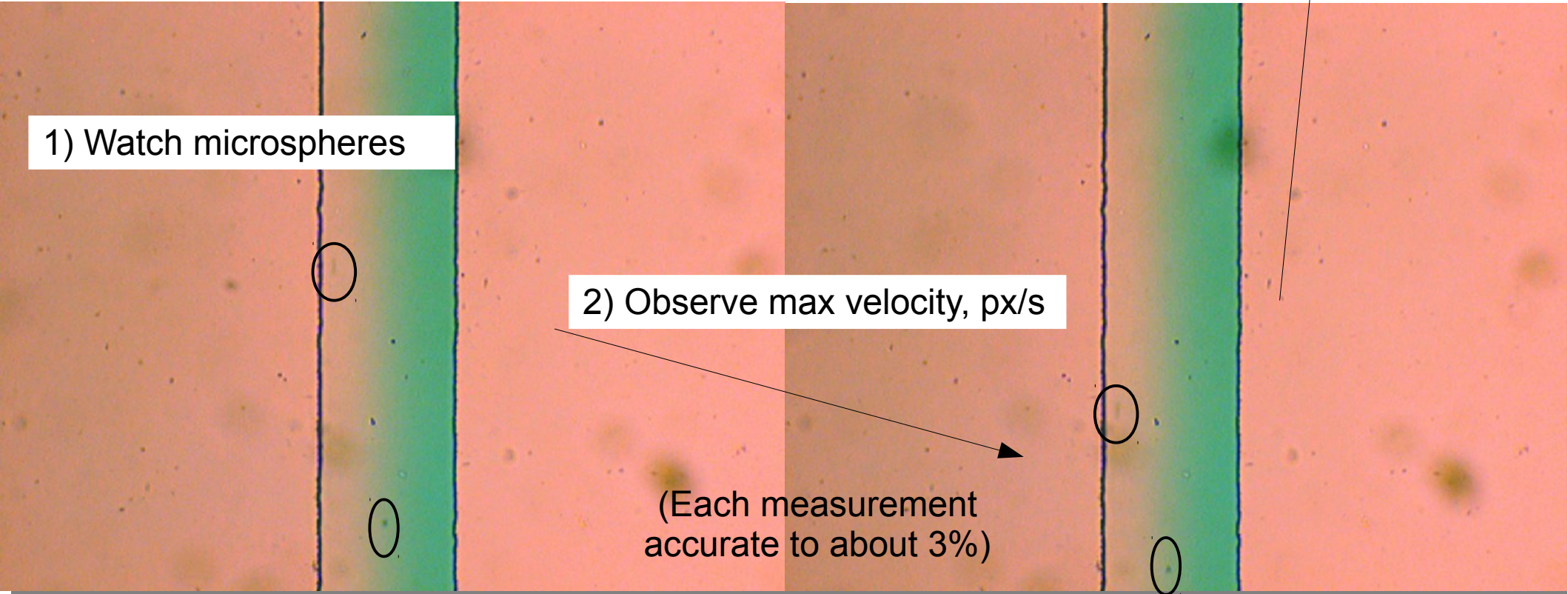
(Determination of maximum relatively robust because of this distribution)



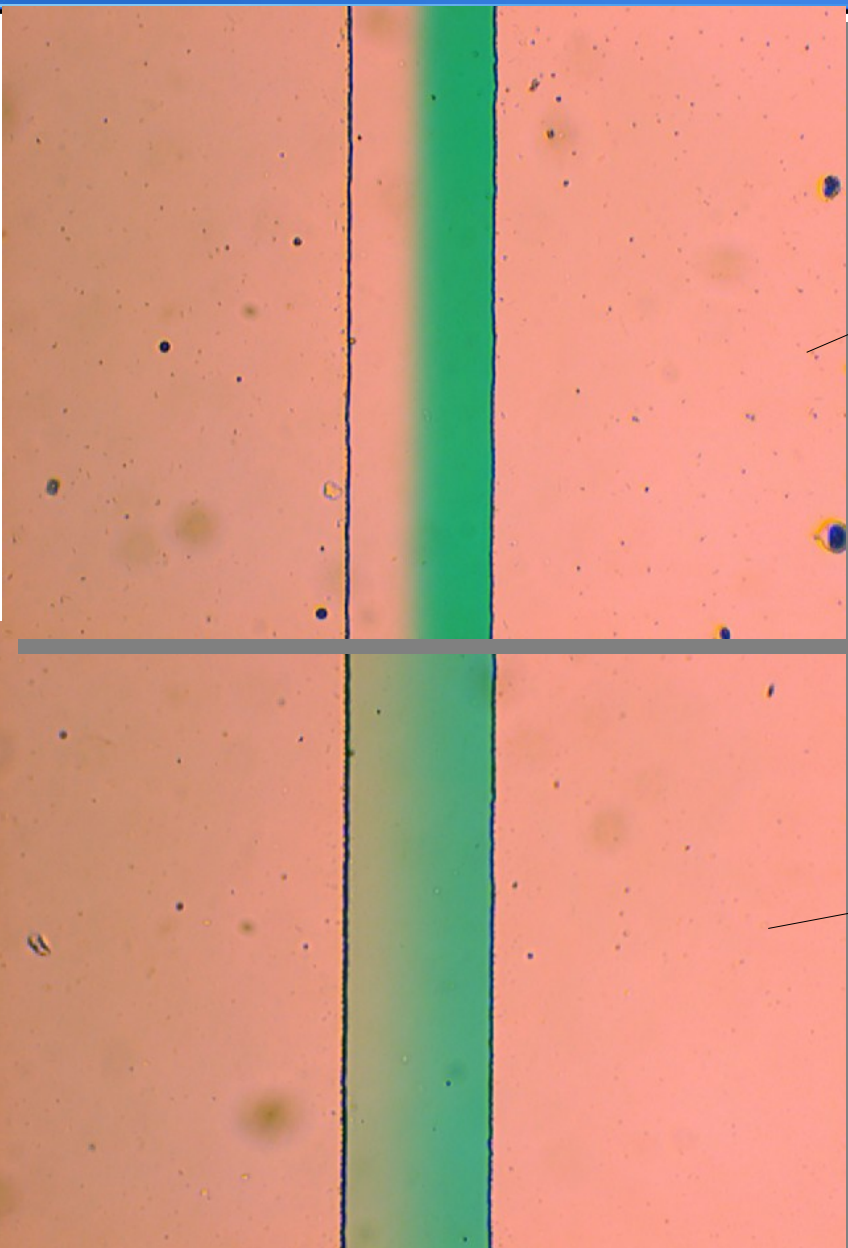
1) Watch microspheres

2) Observe max velocity, px/s

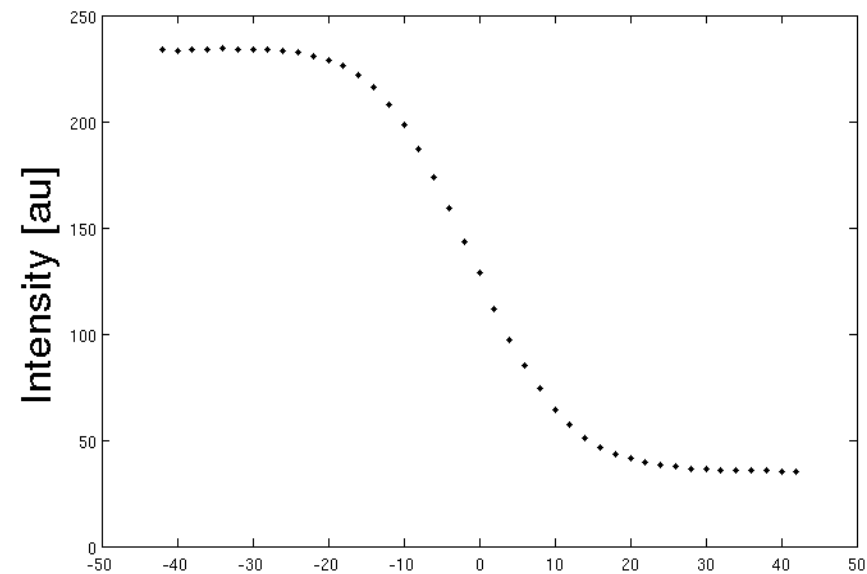
(Each measurement accurate to about 3%)



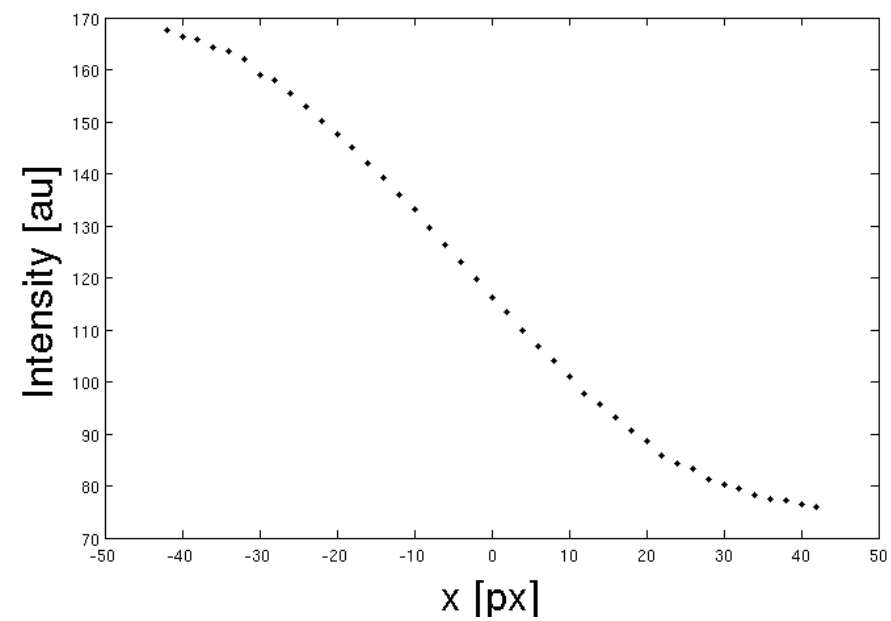
Examine diffusion profile



$z=2\text{mm}$

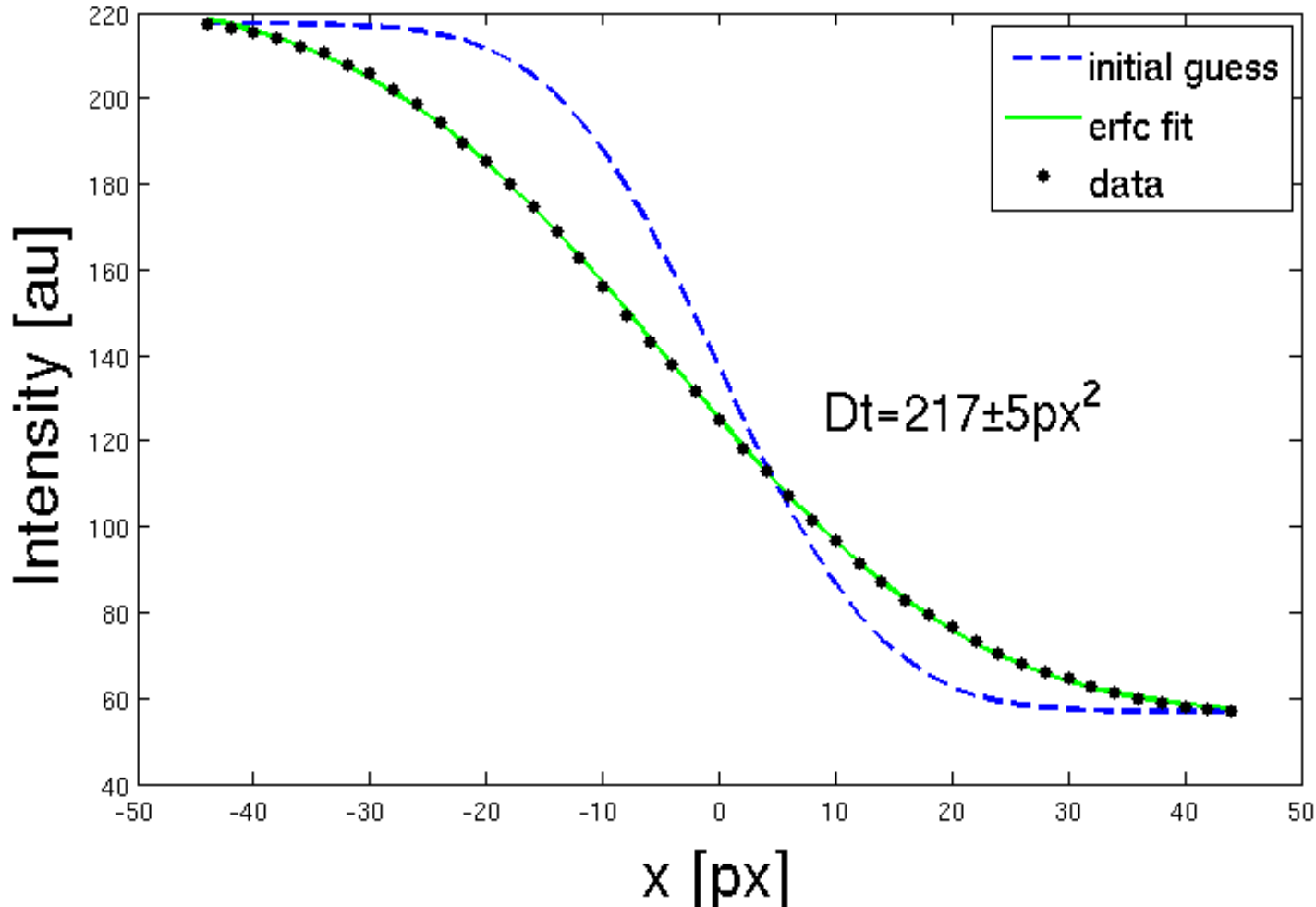


$z=28\text{mm}$



Fit each profile to model

$z=10\text{mm}$



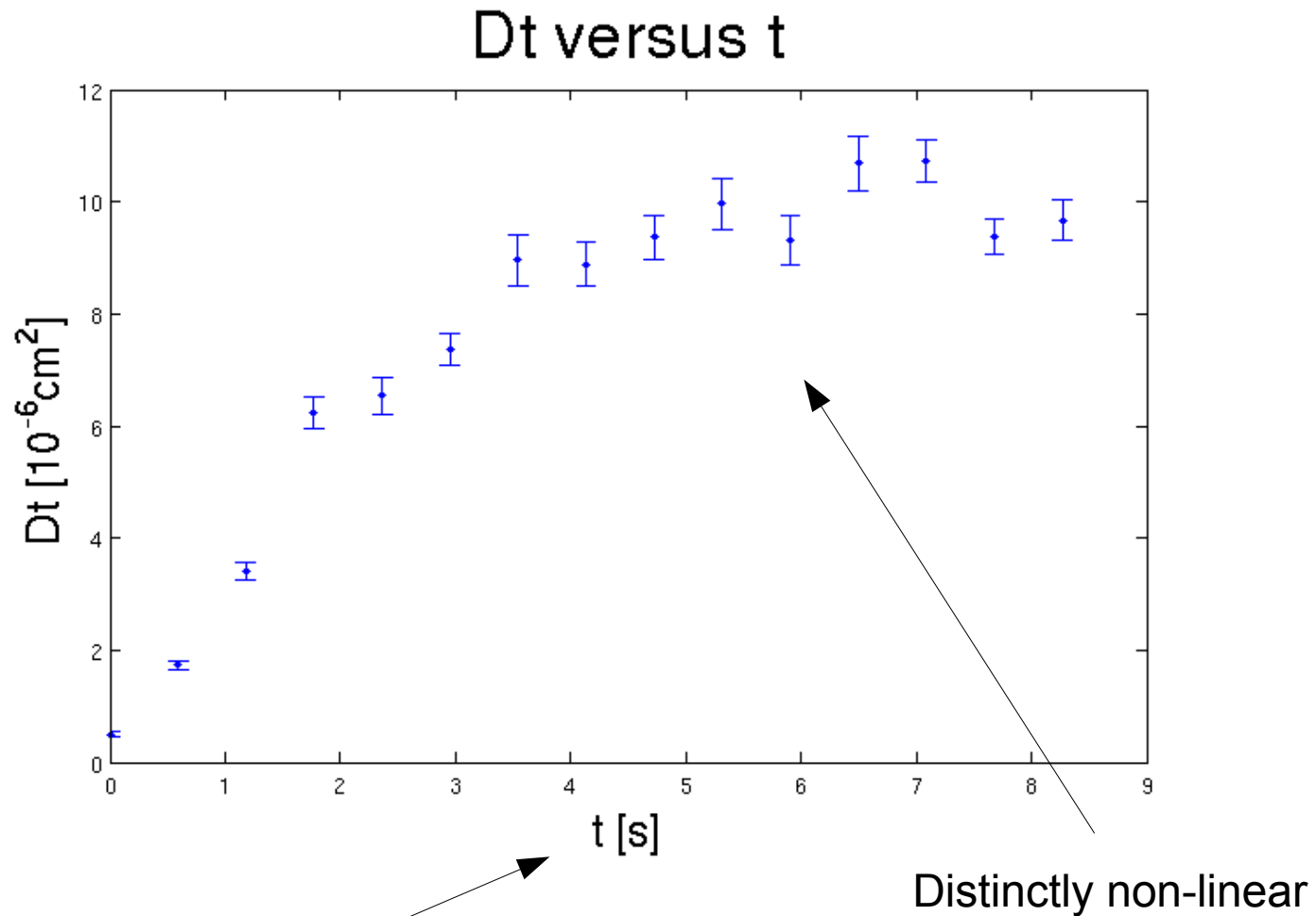
$$I = I_0 - \frac{c}{2} \left(1 + \text{erf} \left(\frac{x-s}{l} \right) \right)$$

From solving
diffusion equation

$$l = 2\sqrt{Dt}$$

So each image
gives us a value
for Dt

Combine estimates

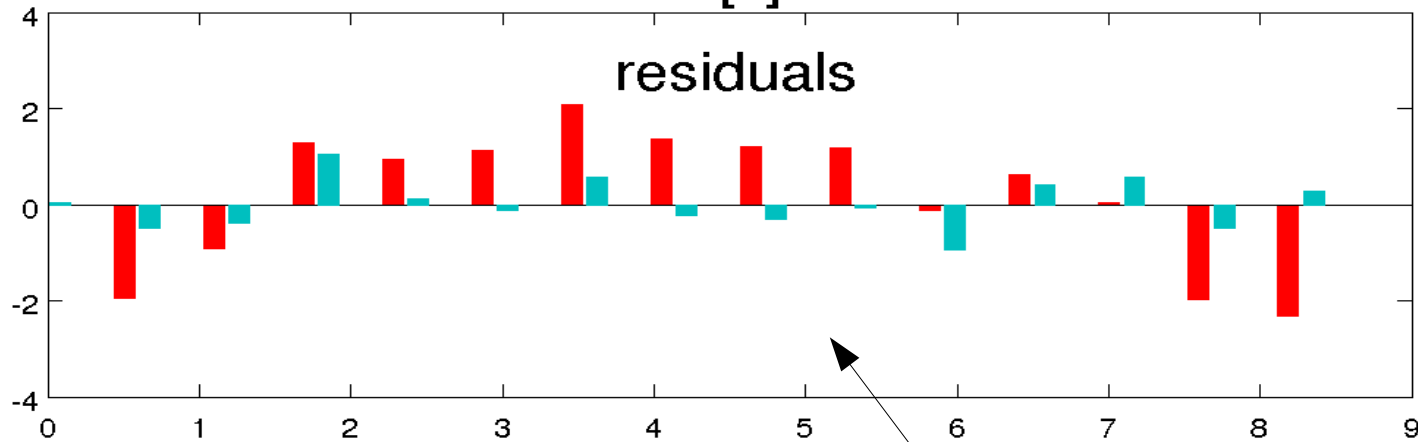
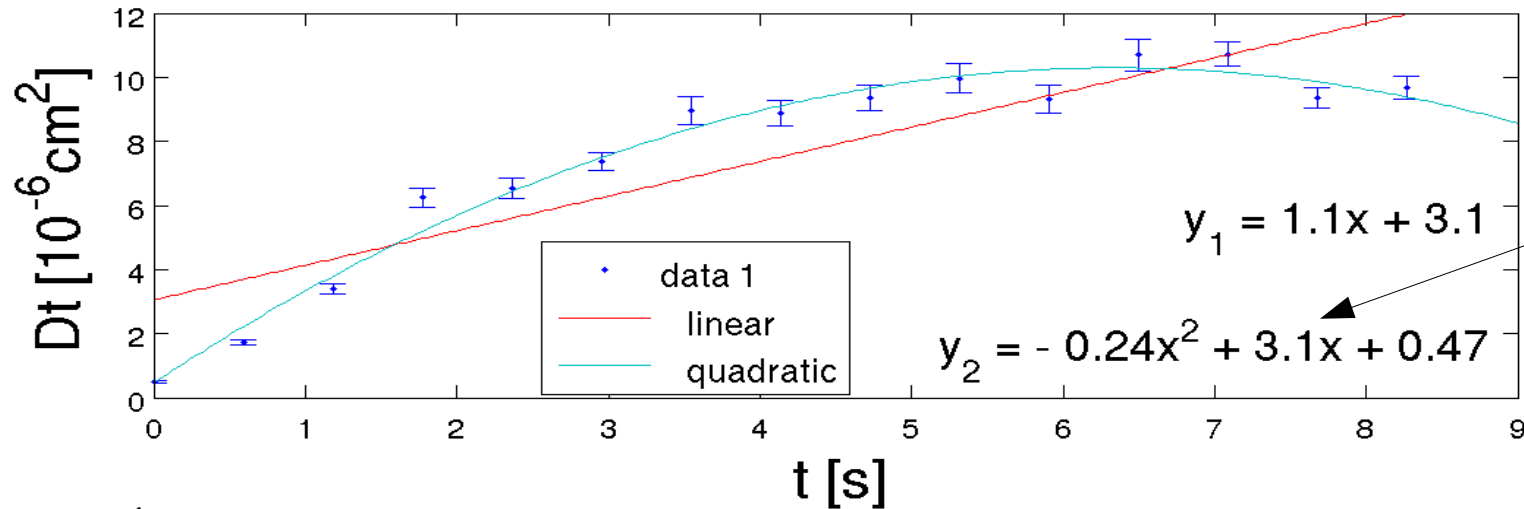


Use measured velocity
to convert z-axis to time

Fit results

In the extent to which a “diffusion constant” is a reasonable parameter for describing this data, it would be this linear-term fit value $D=3.1\text{cm}^2/\text{s}$.

Dt versus t



Quadratic fit empirically justifiable

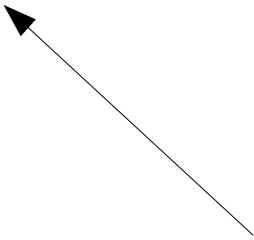
Consider assumptions/ Why is it non-linear?

- Flow velocity is non-uniform
 - Not a 1-D diffusion equation at each cross-section.
- Effect of left boundary down the channel
 - Slows diffusion because no leakage off to infinity
- Absorption is not a linear function of concentration
 - Beer-Lambert law

Other interesting calculations

- Justify choice of flow model with Reynold's number

$$Re = \frac{\rho v_{avg} D_H}{\eta}$$

$$D_H = \frac{4 \text{ Area}}{\text{Perimeter}}$$


Our Reynold's number is .32, which means we don't expect any sort of turbulence.

($Re < 1$ is typical in microfluidics.)

Other interesting calculations

- Further examine of the validity of the model via pressure predictions:

$$P_{model} = \frac{v_{max} \cdot 8 \eta \cdot l}{h}$$

Measured by scope
(to about 3%)

(Easy to measure
by eye within 1%)

$$P_{actual} = \rho g (H_{clear} + H_{dye})$$

Measured with
surface profiler
(constant within
about to 4%)

$$P_{model} = 712 \text{ Pa}$$

$$P_{actual} = 804 \text{ Pa}$$

The model is off by about
11%, which is barely
beyond our error budget.

In summary

- We found a diffusion coefficient for the system, but found that the diffusion did not proceed linearly as our model would have hoped. We explained possible reasons for this discrepancy.
- We also examined other calculations within the model for consistency, and found reasonable Reynold's numbers and appropriate pressures.

Thank you!

Image credits:

- <http://www.thinxxs.com/main/thinxxs/mailling-california.html>
- <http://www.technologyreview.com/news/416410/ibms-move-in-microfluidics/>
- <http://glossary.periodni.com/glossary.php?en=laminar+flow>

